Preparation of 17 β -(Hydroxymethyl)-3-methoxyestra-1,3,5(10)trien-18-oic Acid 18,20-Lactone, a New C(18)-Oxygenated Steroid

by Alexander Kuhl and Wolfgang Kreiser*

Naturstoffchemie, Universität Dortmund, Otto-Hahn-Str. 6, D-44221 Dortmund

Lactone 13 was synthesized by intramolecular radical-mediated oxygenation of the angular Me(18) group of the 17b-configurated sterol 3. Several substrate-controlled methods for stereoselective construction of the hydroxymethyl side chain of 3 were investigated.

Introduction. - Albeit 18-norsteroids appear quite similar to their naturally occurring counterparts $-$ they differ only in the missing Me group $-$ their synthesis has proven a rather difficult endeavour, and is in the case of 18-norestradiol (1) also somewhat ambiguous $[1-3]$. In our attempt for a new approach to the latter compound [4], we reinvestigated the various techniques to functionalize the angular Me(18) group in estradiol (2) itself and in its derivatives. Apart from the known retro-pinacol rearrangement¹) and *Beckmann* fragmentation [6] of the corresponding 17-ketoxime, we examined an intramolecular radical-mediated oxygenation, the well-known hypoiodite reaction [7a]. To meet the configurative prerequisite that non-activated C-atoms in δ -position of alcohols are prone to attack by the *in situ* formed alkoxy radical [8], we had to introduce a β -configurated hydroxyalkyl side chain at C(17) in the case of 2. Since no further additional chiral center at $C(20)$ should be created, we anticipated the hydroxymethyl derivative 3 to be a conceivable precursor for the hypoiodite reaction²).

Results and Discussion. $-$ Sterol 3 has previously been synthesized [9] in a total yield of 8 or 17%, respectively, starting from epoxide 4 which was easily obtained $[10][11]$ (*Scheme 1*). However, this route suffers from the pyrolytic cleavage of 4

¹⁾ Alkene 5 is obtained from estradiol 3-methyl ether or the corresponding 17-(tosyloxy) derivative (see [1] and [5]). In accordance with the literature [1] [5], the $\Delta^{13,14}$ isomer is also obtained in minor amounts (8%). Purification by TLC with AgNO₃-impregnated plates (22.5% AgNO₃, cyclohexane/AcOEt 19:1; R_f 0.60) provides pure 5 in 52 and 50% yield, respectively. Attempts to further functionalize the vinylic methyl group by treatment with SeO₂ lead to a complex mixture of oxygenation products, leaving, however, the vinylic methyl group untouched.

²) A hydroxymethyl group β -situated at C(17) instead of a substituted alkyl side chain has been demonstrated to have a yield-increasing effect in the pregnane series, due to missing steric hindrance [7b].

yielding mainly alkene 5. Our preliminary attempts to prepare 3 from 4 by a Lewisacid-catalyzed reductive ring opening [12] on treatment with $BF_3 \cdot Et_2O$ and $NaBH_3CN$ have led to alcohol 3 with high diastereoselectivity $(a/\beta 5:95)$, but in only 46% yield.

Therefore, two alternative routes have been developed (Scheme 2). The first one, which proceeds under complete diastereoselective control, starts from unsaturated nitrile 7 [13]. The latter is obtained by treatment of estrone methyl ether (6) with 2 equiv. of Me₃SiCN (TMSCN) [14] and subsequent dehydration with dry $Et₄NF/$ $POCl₃$ in abs. pyridine. This one-pot procedure leads $-$ after normal workup, followed by chromatography or *Soxhlet* extraction and crystallization $-$ to pure 7 in 70% yield. The next four steps run very smoothly and in almost quantitative yield. Thus, saponification of 7 with NaOH in ethanol at elevated temperature (195 $^{\circ}$, sealed autoclave) and esterification of the resulting crude acid with $Me₂SO₄$ yields – after crystallization from $MeOH -$ the previously unknown methyl ester 8. Due to shielding by the Me(18) group, hydrogenation of $\bf{8}$ (3 bar, Pd/C in AcOEt/AcOH) occurs exclusively from the α -face, yielding pure 17 β -configurated ester 9, the physical properties of which are in accordance with the available literature data³ $)$ ⁴). Finally, reduction with LiAlH₄ in THF gave the pure alcohol $3⁵$) in 64% yield with respect to ketone 6. The correct β -configuration at C(17) of 3 is checked by 2D NOESY NMR experiments (cross-peak between Me(18) (s at 0.67 ppm) and the two diastereotopic $H-C(20)$ (2dd at 3.46 – 3.59 and 3.71 – 3.76 ppm)).

The second, even shorter, but in comparison with the above mentioned sequence lower-yielding route provides 3 via alkene 10 [16]. The latter is obtained from 6 on treatment with 4 equiv. of triphenylphosphonium methylid in 78% yield and then submitted to hydroboration and subsequent oxidative workup. The use of diborane in THF at 0° leads in our hands [9b] to a mixture of 3 and the undesirable $17a$ diastereoisomer in a ratio of $3:1$ (yield 62%), as indicated by ¹³C-NMR spectroscopy⁶). Best results are achieved with 9-borabicyclo[3.3.1]nonane (9-BBN) at 0° , leading after oxidative workup and column chromatography to 3 and only traces of the

³⁾ Hydrogenation of 7 is unfavourable since prolonged reaction times are needed. Furthermore, the saponification of the corresponding saturated nitrile leads to a 3:1 mixture, with preference for the 17β diastereoisomer.

⁴⁾ In the course of the structure elucidation of ouabagenine, a cardenolide from Acokanthera schimperi, etianic acid derivative 9 was prepared by hydrogenation of the corresponding \mathcal{A}^{14} compound, a degradation product [15].

⁵⁾ Our sample of 3 shows a m.p. identical with the one given in [9a], whereas its optical rotation agrees with [9b] (see *Exper. Part*).

⁶) Determined by integration of the ¹³C-NMR signals of Me(18) of the 17 β -isomer 3 and its 17 α -isomer at 12.5 and 20.7 ppm, respectively.

corresponding epimer (α/β 5:95). The latter can be separated by a single crystallization from MeOH, yielding pure $3⁵$) in 46% with respect to 6. It should be noted that these results do not differ from those achieved by direct application of in situ prepared 9- BBN. Since the latter is simply formed by reaction of inexpensive cycloocta-1,5-diene with NaBH₄ and BF₃ Et₂O, this method can be recommended as an economic alternative to the known procedure [17] utilizing neat 9-BBN.

The functionalization of the angular Me(18) group succeeds through a sequence of oxygenation steps (Scheme 3). Thus, a suspension of 3, lead tetraacetate, powdered $CaCO₃$ (as a buffer), and iodine in cyclohexane is heated under reflux to give the primary products of the hypoiodite reaction [7a], the tetrahydro-2-iodofurans 11 which are not isolated. Instead the crude reaction mixture is heated in the presence of a solution of NaOAc in AcOH, providing the acylated hemiacetals $12⁷$ by displacement of the I-atom. These are directly oxidized [18] with CrO_3 pyridine complex [19] to afford an oil (73%) from which pure lactone 13 precipitates after trituration with $Et₂O.$

The structure of lactone 13 is unequivoqually assigned by $H-MMR$ spectroscopy and independantly established by X-ray diffraction (see Fig.) [20]⁸); there is agreement with the spectroscopic data (see Table and Exper. Part).

⁷) The ¹H-NMR spectrum of crude 12 shows the acetoxy groups at 2.05 (s) and 2.07 (s) ppm.

⁸⁾ Observed torsion angles φ : H_{si} - C(20) – C(17) – H 88.7(5)°; H_{re} - C(20) – C(17) – H 32.5(5)°.

The oxymethylene group of the lactone moiety of 13 shows a significant set of ¹H-NMR signals. In contrast to the dd of the re located H $-C(20)$ (4.35 - 4.38 ppm), only a d (4.12 ppm) is observed for the si facial H $-C(20)$, which can be explained by means of a perpendicular orientation relative to the vicinal $H-C(17)$ on application of the Karplus relation [21].

Figure. Molecular structure of 13

Experimental Part

General. All reagents and solvents are commercially available and used without further purification. Abs. CH_2Cl_2 is freshly distilled from 4- \AA molecular sieve. Usual workup includes the washing of the org. layer with a sat. NaHCO₃ soln. and brine, followed by drying (Na₂SO₄ or MgSO₄), evaporation, and drying in vacuo. TLC: Merck silica gel 60 F_{254} plates (Art. No. 5554); detection by UV, I₂, and phosphomolybdic acid. Column chromatography (CC): columns of different size filled with silica gel 60 (230 - 400 mesh) from Merck Co. M.p.'s: instrument model 571 by Büchi Co; not corrected. Optical rotations: Perkin-Elmer-141 automatic polarimeter (10-cm, 1-ml cell); at r.t. IR Spectra: *Nicolet-Impact-400* instrument; films or KBr discs in cm⁻¹. NMR Spectra: *Bruker DRX 400* (¹H, 400.13 MHz; ¹³C 100.61 MHz); δ in ppm downfield from internal Me₄Si (= 0 ppm) and

CDCl₃ (77.0 ppm), J in Hz. MS: MAT 8230 by Finnigan Co. (EI 70 eV); m/z (rel. intensities in %). The elemental analyses were carried out on a Leco-CHNS-932 instrument.

 $3-Methoxyestra-1,3,5(10),16-tetraene-17-carbonitrile (7)$. A soln. of 6 (1.42 g, 5.0 mmol) in abs. CH₂Cl₂ (20 ml) is treated with Me₃SiCN (0.99 g) and ZnI₂ (0.05 g) for 72 h under reflux. Evaporation yields the 17 β protected cyanohydrine as a single diastereoisomer which is dissolved in abs. pyridine (6 ml) and added dropwise at r.t. within 20 min to a soln. of $Et₄NF (1.19 g, 8.0 mmol)$ and POCl₃ (6.1 g, 40 mmol) in abs. pyridine (11 ml). After stirring for 60 min at r.t. and 90 min under reflux, the cooled soln. is poured onto ice $(100 g)$. Acidification with conc. HCl soln., separation of the precipitate, and usual workup yields crude 7 which is purified by Soxhlet extraction Et₂O (425 ml) CH₂Cl₂ (25 ml) and crystallization from Et₂O (20 ml) or CC (cyclohexane/AcOEt 4:1; R_f 0.47): 1.04 g (70%) of 7. M.p. 166–168°. [α] $^{20}_{D}$ = + 68 (c = 1.00, CHCl₃). IR (KBr): 2210 (CN). ¹H-NMR (400 MHz, CDCl₃): 0.94 (s, Me(18)); 2.88–2.92 (m, 2 H–C(6)); 3.77 (s, MeO); 6.64 $(d, {}^{4}J(2,4) = 2.5, H - C(4)); 6.65 - 6.66 (dd, {}^{3}J(16,15) = 1.9, {}^{3}J(16,15) = 3.0, H - C(16)); 6.70 - 6.73 (dd, {}^{3}J(2,1) =$ $8.5, \frac{4J(2,4) = 2.6, \text{ H} - \text{C}(2))$; 7.25 $(d, \frac{3J(1,2) = 8.5, \text{ H} - \text{C}(1))}{2}$. ¹³C-NMR: *Table.* MS: 293 (100, M⁺⁺), 278(6), 186 (12), 173 (17), 160 (67), 43 (95). Anal. calc. for C₂₀H₂₃NO: C 81.87, H 7.90, N 4.77; found: C 81.30, H 8.20, N 4.75.

Methyl 3-Methoxyestra-1,3,5(10),16-tetraene-17-carboxylate (8). In a sealed autoclave (200 ml), 7 (0.59 g, 2.0 mmol) and a soln. of NaOH (2.0 g, 50 mmol) in H₂O (10 ml) are heated for 9 h at 195 $^{\circ}$. Evaporation and acidification with conc. HCl soln. (8.0 ml) yield after filtration, the crude acid, which is treated with Me₂SO₄ (0.39 ml) and K_2CO_3 (0.69 g) abs. acetone (75 ml) under reflux for 24 h. The ester obtained after usual workup is purified by crystallization from MeOH (40 ml): 0.61 g (95%) of **8**. M.p. $136 - 138^{\circ}$. $[\alpha]_D^{20} = +82$ ($c = 1.00$, CHCl₃). IR (KBr): 1709 (CO). ¹H-NMR (400 MHz, CDCl₃): 0.95 (s, Me(18)); 2.85 – 2.92 (m, 2 H – C(6)); 3.74 $(s, COOMe)$; 3.77 (s, MeO) ; 6.63 $(d, 4J(2,4) = 2.7, H-C(4))$; 6.70 – 6.72 $(dd, 3J = 8.6, 4J(2,4) = 2.7, H-C(2))$; 6.79 – 6.81 $(dd, {}^3J(16,15) = 1.8, {}^3J(16,15) = 3.3, H - C(16)$; 7.20 $(d, {}^3J(1,2) = 8.6, H - C(1))$. ¹³C-NMR: *Table.* MS: 326 (100, M⁺), 311 (11), 295 (10), 267 (10), 198 (21), 173 (19). Anal. calc. for C₂₁H₂₆O₃: C 77.27, H 8.03; found: C 77.60, H 8.20.

Methyl 3-Methoxyestra-1,3,5(10)-triene-17-carboxylate (9). A soln. of 8 (0.494 g, 1.50 mmol) in AcOEt (150 ml) and AcOH (0.3 ml) is shaken for 28 h with 10% Pd/C (0.1 g) under 3 bar H_2 pressure. Usual workup gives pure 9 (0.483 g, 98%). M.p. 165–166°. [α] $_{\text{D}}^{\text{20}}$ =+97 (c =1.00, CHCl₃). IR (KBr): 1730 (CO). ¹H-NMR $(400 \text{ MHz}, \text{ CDCl}_3): 0.69 \text{ (s, Me(18))}; 2.44 \text{ (t, } 3(17,16) = 9.1, \text{ H}-\text{C}(17)); 2.84 - 2.86 \text{ (m, 2 H}-\text{C}(6)); 3.69 \text{ (m, 3 H}-\text{C}(6)); 3.69 \text{ (m, 4 H})$ (s, COOMe); 3.77 (s, MeO); 6.62 $(d, {}^{4}J(2,4) = 2.6, H - C(4))$; 6.69 – 6.72 $(dd, {}^{3}J(2,1) = 8.6, {}^{4}J(2,4) = 2.6$, $H-C(2)$; 7.20 $(d, {}^{3}J(1,2) = 8.6, H-C(1))$. ¹³C-NMR: *Table*. MS: 328 (100, M⁺⁺), 297(6), 269 (4), 227 (16), 199 (51), 173 (44). Anal. calc. for $C_{21}H_{28}O_3$: C 76.79, H 8.59; found: C 76.10, H 8.85.

Methoxyestra-1,3,5(10)-triene-17 β -methanol (3). a) From 9. A soln. of 9 (0.443 g, 1.35 mmol) in abs. THF (25 ml) is added within 5 min to a suspension of LiAlH₄ $(0.103 \text{ g}, 2.7 \text{ mmol})$ in abs. THF (30 ml) at r.t. The mixture is stirred for 24 h 0° , then treated with H₂O (0.3 ml) and 5% NaOH soln. (0.6 ml) and stirred for 12 h at r.t. Then the mixture is dried (Na₂SO₄) and evaporated: pure 3 (0.398 g, 98%). M.p. 108 - 110° ([9a]: 108 - 109°; [9b]: 125°). [α] $_{\text{D}}^{\text{20}}$ = + 62 (c = 0.99, CHCl₃) ([9a]: [α] $_{\text{D}}^{\text{20}}$ = + 23 (c = 0.50, CHCl₃); [9b]: [α] $_{\text{D}}^{\text{20}}$ = + 64.5). IR (KBr): 3450 (OH). ¹H-NMR (400 MHz, CDCl₃): 0.67 (s, Me(18)); 2.83–2.86 (m, 2 H–C(6)); 3.46 (s, OH); 3.46 $(dd, {}^{2}J(20,20') = 10.5, {}^{3}J(20,17) = 7.5, H-C(20)$; 3.71 – 3.76 $(dd, {}^{2}J(20,20') = 10.5, {}^{3}J(20,17) = 6.8, H-C(20)$; 3.76 (s, MeO); 6.62 $(d, {}^{4}J(2,4) = 2.7, H-C(4))$; 6.68 – 6.72 $(dd, {}^{3}J(2,1) = 8.6, {}^{4}J(2,4) = 2.7, H-C(2))$; 7.19 $(d, {}^{3}J(1,2) = 8.6, H - C(1)).$ ¹³C-NMR: Table. MS: 300 (100, M⁺⁺), 227 (7), 199 (26), 186 (12), 173 (31), 160 (22). Anal. calc. for $C_{20}H_{28}O_2$: C 79.95, H 9.39; found: C 80.35, H 9.20.

b) From 10. To a suspension of NaBH₄ (0.48 g) in abs. 1,2-dimethoxyethane (10 ml), THF (50 ml) and cycloocta-1,5-diene (1.73 g, 16.0 mmol) is added dropwise $BF_3 \cdot Et_2O$ (2.41 g). After 2 h stirring at 0° and 2 h under reflux, a soln. of 10 (2.26 g, 8.0 mmol) in abs. THF (25 ml) is added within 5 min at 0° . The mixture is stirred for 40 h at 0° and then treated with 30% H₂O soln. (8.0 ml) and 10% NaOH soln. (8.0 ml) for 4 h at 0° . Quenching with conc. NH₄Cl soln. (50 ml) followed by repeated extraction with Et₂O (450 ml; in total) gives, after usual workup, 4.0 g of an oil which is purified by CC (cyclohexane/AcOEt 1:1; R_f 0.53). Crystallization from MeOH (5.0 ml) yields pure 3 (1.41 g, 59%). This material is identical to that obtained above.

 $3-Methoxy-17-methylen-estra-1,3,5(10)-triene$ (10). A soln. of 6 (2.84 g, 10.0 mmol) abs. THF (250 ml) is added under reflux within 1 h to a soln. of the methylid prepared from methyltriphenylphosphonium bromide (14.3 g, 40 mmol) in abs. THF (150 ml) and 2.5m BuLi in hexane (18 ml, 45 mmol). After 7 h under reflux, the cooled mixture is treated with $H₂O (100 ml)$ and evaporated. Acidification with dil. $H₂SO₄$ soln. and extraction with CHCl₃ (200 ml) yields, after usual workup crude 10, which is purified by CC (cyclohexane/AcOEt 2:1; R_f 0.76) and crystallization from MeOH: **10** (2.19 g, 78%). M.p. 80–81°. [α] $_{10}^{20}$ = + 58 (c = 1.04, CHCl₃). IR (KBr): 1653 (C=C). ¹H-NMR (400 MHz, CDCl₃): 0.81 (s, Me(18)); 2.84 – 2.89 (m, 2 H–C(6)); 3.76 (s, MeO);

4.67 (br. s, 2 H – C(20)); 6.63 (d, $\mathcal{U}(2,4) = 2.5$, H – C(4)); 6.70 – 6.73 (dd, $\mathcal{U}(2,1) = 8.7$, $\mathcal{U}(2,4) = 2.5$, H – C(2)); 7.25 $(d, {}^{3}J(1,2) = 8.7, H-C(1))$. ¹³C-NMR: *Table*. MS: 282 (100, M⁺⁺), 267 (3), 239 (3), 225 (19), 186 (23), 173 (32). Anal. calc. for $C_{20}H_{26}O$: C 85.05, H 9.28; found: C 85.15, H 9.15.

17b-(Hydroxymethyl)-3-methoxyestra-1,3,5(10)-trien-18-oic Acid 18,20-Lactone (13). To a suspension of finely powdered Pb(OAc)₄ (3 g, 6.8 mmol) and CaCO₃ (1 g), obtained after 5 min stirring under reflux in cyclohexane (100 ml) , I₂ $(0.86 \text{ g}, 3.4 \text{ mmol})$ is added in a single batch. After 1 h boiling, 3 $(0.398 \text{ g}, 1.32 \text{ mmol})$ is added neat to the hot suspension. Heating under reflux is continued for 4 h and then the slightly yellow suspension filtered hot. The filtrate is washed with sat. $\text{Na}_2\text{S}_2\text{O}_3$ soln. $(2\times10\,\text{ml})$ and evaporated. The obtained colourless oil is mixed with AcOH (12 ml) , H₂O (3 ml) and NaOAc (1.5 g) and heated for 3 h under reflux. The cooled soln. is neutralized with NaHCO₃ soln. (100 ml) and extracted with $Et_2O(3 \times 50 \text{ ml})$. The solvent is evaporated and the residue dissolved at 0° in a soln. of CrO₃ (0.74 g) in H₂O/pyridine 1:1 (3 ml). After stirring for 24 h, the mixture is combined with an ice-cold soln. of NaHSO₃ (2 g) in H₂O (15 ml). Acidification with 1n HCl, followed by extraction with Et₂O (150 ml) yields, after usual workup, 0.30 g (73%) of impure **13**. An anal. sample (58 mg) is obtained by trituration with Et₂O (3.0 ml). M.p. 191 – 192°. [α] $^{24}_{\rm D}$ = + 39 (c = 1.00, CHCl₃). IR (KBr): 1750 (CO). ¹H-NMR (400 MHz, CDCl₃): 2.79 – 2.94 $(m, 2 H - C(6))$; 3.76 (s, MeO); 4.12 $(d, {}^{2}J(20si, 20re) = 9.1, H_{si} - C(20))$; 4.35 – 4.38 $(1dd, {}^{2}J(20re, 20si) = 9.1, {}^{3}J(20re, 17) = 5.2, H_{re} - C(20))$; 6.62 $(d, {}^{4}J(2,4) = 2.5, H-C(4));$ 6.68 - 6.71 $(dd, {}^{3}J(2,1) = 8.6, {}^{4}J(2,4) = 2.5, H-C(2));$ 7.19 $(d, {}^{3}J(1,2) = 8.6,$ $H-C(1)$). ¹³C-NMR: *Table*. MS: 312 (100, M⁺⁺), 298(8), 284(8), 253(4), 225(3), 199(8), 191(17). Anal. calc. for $C_{20}H_{24}O_3$: C 76.89, H 7.74; found: C 77.15, H 7.90.

REFERENCES

- [1] W. F. Johns, J. Org. Chem. 1961, 26, 4583.
- [2] W. F. Johns, J. Org. Chem. 1963, 28, 1856.
- [3] W. F. Johns, *J. Org. Chem.* **1968**, 33, 109.
- [4] A. Kuhl, H. Karels, W. Kreiser, submitted to Helv. Chim. Acta.
- [5] G. Stork, H. N. Khastgir, A. J. Solo, *J. Am. Chem. Soc.* **1958**, 80, 6457.
- [6] A. H. Fenselau, E. H. Hamamura, J. G. Moffat, J. Org. Chem. 1970, 35, 3546.
- [7] a) J. Kalvoda, K. Heusler, Synthesis 1971, 501; b) ibid. 1971, 522 and 523 (Table 7).
- [8] D. H. R. Barton, J. M. Beaton, L. E. Geller, M. M. Pechet, J. Am. Chem. Soc. 1960, 82, 2640.
- [9] a) U. Klein, W. Sucrow, Chem. Ber. 1977, 10, 2401; U. Klein, Ph.D. Thesis, Technical University Berlin, 1975; b) Roussel-Uclaf, Pat., Fr 2332759 (CA: 1978, 88, 121538).
- [10] E. J. Corey, M. Chaykovsky, J. Am. Chem. Soc. 1965, 87, 1353.
- [11] M. Hübner, I. Noack, *J. Prakt. Chem.* **1972**, 314, 667.
- [12] R. O. Hutchins, I. M. Taffer, W. Burgoyne, J. Org. Chem. 1981, 46, 5214.
- [13] D. Burn, V. Petrow, J. Chem. Soc. 1962, 364.
- [14] P. G. Gassman, J. J. Talley, *Tetrahedron Lett*. **1978**, *19*, 3773.
- [15] Ch. Tamm, G. Volpp, G. Baumgartner, Helv. Chim. Acta 1957, 40, 1469.
- [16] R. Kanojia, S. Rovinsky, I. Scheer, J. Chem. Soc., Chem. Commun. 1971, 1581.
- [17] H. C. Brown, E. F. Knights, C. G. Scouten, J. Am. Chem. Soc. 1974, 96, 7765.
- [18] C. Meystre, K. Heusler, J. Kalvoda, P. Wieland, G. Anner, A. Wettstein, Helv. Chim. Acta 1962, 45, 1317.
- [19] G. I. Poos, G. E. Arth, R. E. Beyler, L. H. Sarett, J. Am. Chem. Soc. 1953, 75, 422.
- [20] A. Kuhl, A. Kornath, M. Schürmann, H. Preut, W. Kreiser, Acta Crystallogr., Sect. C 1998, 54, 1115.
- [21] M. Karplus, *J. Am. Chem. Soc.* **1963**, 85, 2870.

Received August 7, 1998